

### **General Chemistry Laboratory**

# Synthesis of Superconductor

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Last revised: 2025/02/06



**Preparation** 

### **Collect the following items**

 Agate mortar (clean with sponge after use)
Label one zip-lock bag with student ID and name (To collect the synthesized superconductor)

### From your personal equipment

- One plastic spatula
- Mask (self-prepared)





# **Superconductors**

### Metal conductors

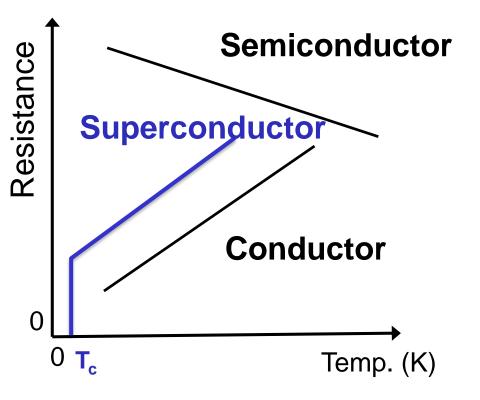
Conductivity decreases with increasing temperature

### Semiconductors

Conductivity increases with increasing temperature

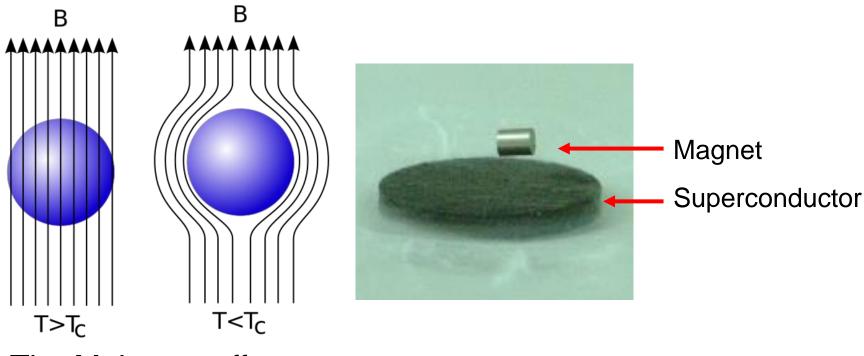
### Superconductors

Zero resistance at transition temperature,  $\rm T_{\rm c}$ 





# **Properties of Superconductor**



The Meissner effect

- At superconducting transition temperature, T<sub>c</sub>
  - Zero resistance
  - Meissner effect



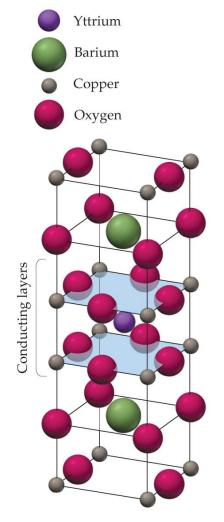
# **YBCO Superconductor**

- Yttrium Barium Copper Oxide (YBCO) superconductor
- Chemical formula: YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-x</sub>
- Crystalline structure

# T<sub>c</sub>: 95 K (Boiling p)

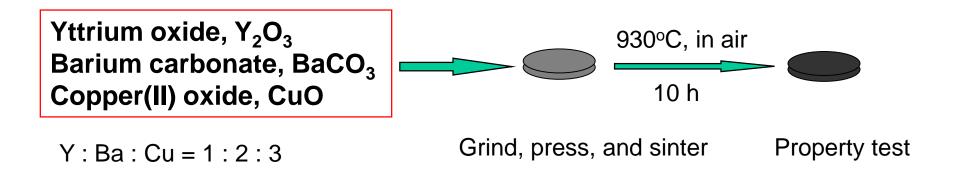
(Boiling point of liquid nitrogen: 77 K)

- Synthetic methods
  - Solid-state reaction
  - Sol-gel method
  - Co-precipitation





# **Experiment Tasks**



Synthesize YBCO superconductor by solid-state reaction

- 1. Weigh starting materials according to atomic ratio
- 2. Grind the chemicals to homogeneous
- 3. Press the mixture into pellet
- 4. Sinter at 930°C for 10 h
- 5. Test the Meissner effect



# **Step 1: Calculation of Formula**

- To synthesize 0.004 mol YBa<sub>2</sub>Cu<sub>3</sub>O<sub>7-x</sub> (Y : Ba : Cu = 1 : 2 : 3)
  - $Y_2O_3$  1/2 × 0.004 × 225.82 = 0.4516  $\Rightarrow$  0.45 g
  - BaCO<sub>3</sub> 2 × 0.004 × 197.31 = 1.5785 ⇒ 1.58 g
  - CuO 3 × 0.004 × 79.55 = 0.9546 ➡ 0.95 g

The stochiometric amount of starting materials is critical to the success of the experiment!



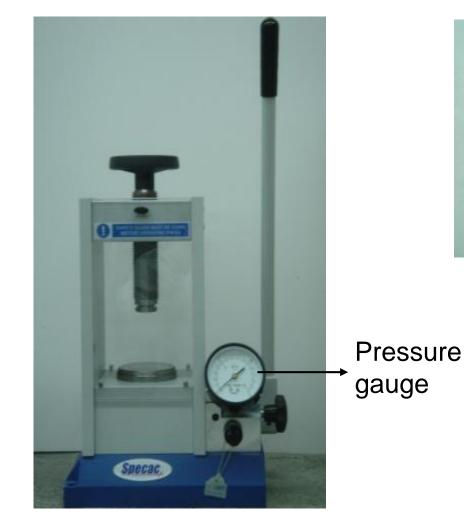


**T** Still contains white powder

- Use agate mortar to contain the chemicals
- Mix chemicals with plastic spatula first, then grind
- Grind until color appears gray and homogeneous, that may take about 10~15 min.
- Collect the ground powder in a piece of weighing paper
- ✓ Agate mortar is expensive (NTD 9,000~15,000) and easily broken, handle with cares



# **Step 3.1: Pressing Apparatus**





Parts of dies

Hollow ring

### Hydraulic press



# **Step 3.2: Press into Pellet**



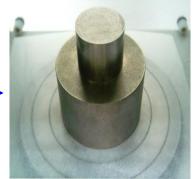


Place die on a weighing paper

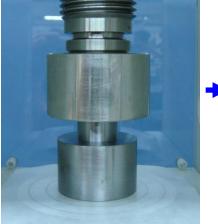
Transfer powder into die evenly



Rotate the die to even up powder



Place the dies on center of hydraulic platform









Release the valve counterclockwise

10

Fix the dies in position

Close the valve clockwise

Apply the **pressure to** 1 ton/cm<sup>2</sup> for 1 min.



# **Step 3.3: Obtain Pellet**

Leveling surface Hollow ring Dies with sample Cylinder

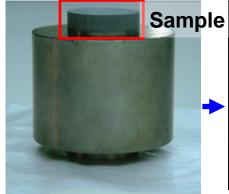
- Place the dies in center of platform upside down
- Setup the hollow ring and leveling surface



- Close the valve clockwise
- Pull leverage to surface the superconductor pellet



- Release the valve
- Take out the hollow ring



Resulting sample pellet





Place pellet onto an alumina plate with plastic tweezers carefully



# Step 4: Sinter at 930°C





- Place the sample pellet into box furnace
- Sinter the samples at 930°C for 10 h. by raising and lowering the temperature at a rate of 5 °C/min.
- ✓ The box furnace can hold  $4 \times 4 = 16$  samples
- Alumina plate cannot be labeled, record the position in furnace instead



# **Step 5: Property Test**





- Examine the Meissner Effect by observing the levitation of magnets
  - Place superconductor in petri dish
  - Place magnetic bars on superconductor pellet with plastic tweezers, compare levitation height and amount of magnetic bars
  - Add liquid nitrogen and immerse the superconductor
  - Dry and warm the superconductor with a hair dryer to room temp.
    - ✓ Notice that the magnetic bars are small and easily lost



- The dies should be placed in center of platform to prevent uneven application of pressure when pressing
- > Pressure should be kept at 1 ton/cm<sup>2</sup> for 1 min.
- Clean up the dies thoroughly after each use to avoid the powders pile up in dies that cause it to get stuck
- Avoid direct contact with liquid nitrogen which may cause frostbite



- Use sponge to clean agate mortar and pressing dies
- Tuck the lab stools underneath the lab bench
- Clean up the lab bench and check personal equipment inventory (have an associate TA signed the check list)
- This is a **Brief Report** experiment:
  - Hand in prelab/lab note/report together to the TA
- Groups on duty shall stay and help clean up the lab